LOYOLA COLLEGE (AUTONOMOUS), CHENNAI - 600 034

B.Sc. DEGREE EXAMINATION - **CHEMISTRY**

SECOND SEMESTER - APRIL 2016

CH 2504/CH 2502/CH 2500 - HYDROCARBONS AND STEREOCHEMISTRY

Date: 21-04-2016	Dept. No.	Max. : 100 Marks
Time: 01:00-04:00		ı

PART-A

Answer ALL Questions

(10x2=20 marks)

- 1. State Huckel's rule.
- 2. What are long lived free radicals? Give an example
- 3. Why is cyclohexane more resistant to ring opening than cyclopropane?
- 4. Predict the product in the following reaction.

- 5. State Hoffmann's rule.
- 6. What happens when acetylene is treated with Ni (CN) 2 in ether?
- 7. Nitration of nitrobenzene is more difficult than nitration of benzene- Why?
- 8. Write the uses of naphthalene.
- 9. Define the term dihedral angle.
- 10. Why chair form of cyclohexane is more stable than its boat form?

PART-B

Answer any EIGHT Questions

(8x5=40 marks)

- 11. Discuss the hybridization involved in ethylene molecule.
- 12. Establish the acid catalyzed mechanism of keto-enol tautomerism.
- 13. Write the mechanism of chlorination of methane.
- 14. How will you prepare cyclopentanone using Dieckmann's ring closure reaction?
- 15. Account for the following:

Addition of HBr to propene (i) in the absence of peroxide yields 2-Bromopropane (ii) in the presence of peroxide yields n-propylbromide.

- 16. Explain the mechanism of hydroboration oxidation of 2-methylpropene.
- 17. Write a note on Ziegler Natta polymerization with an example.
- 18. How naphthalene is synthesized using Haworth's method?

19. How will you effect the following conversion? Write its mechanism.

20. Give the E/Z notation for the following compounds.

- 21. Discuss the conformational analysis of n-butane.
- 22. Write a note on 1, 2 and 1,3 –interactions in substituted cyclohexanes.

PART-C

Answer any FOUR Questions

(4x10=40 marks)

- 23. a) Arrange the following acids in the increasing order of acidity and explain the reason for your answer. **(5)**
 - Monochloro acetic acid, acetic acid, formic acid, propionic acid.
 - b) What is resonance? Explain its conditions and consequences. **(5)**
- 24. Explain the following in detail

(5+5)

- (a) Refining of petroleum (b) Cracking of hydrocarbon.
- 25. Write the products and the reactions when 1-Propyne is treated with
 - a) CH₃COOH b) H₂O/H⁺ c) BH₃

- d) C₂H₅OH
- **(5)**
- 26. a) What is NBS? Explain its usefulness in bromination reaction. b) Write any five important chemical reactions of anthracene.

(5)

27. Account for the following:

(3+3+4)

- a) Only anhydrous AlCl₃ is used in Friedel-Crafts reactions
- b) Pulverized AlCl₃ reduces the reaction time rather than bulk amount.
- c) Friedel-Crafts alkylation reaction leads to polyalkylation
- 28. How are the following methods used to distinguish the geometrical isomers? (10)
 - (a) Melting point (b) Dipole moment (c) Dehydration (d) cyclization (e) Heat of hydrogenation.

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