# LOYOLA COLLEGE (AUTONOMOUS), CHENNAI - 600 034



#### M.Sc. DEGREE EXAMINATION - CHEMISTRY

## THIRD SEMESTER - APRIL 2019

#### CH 3813- COORDINATION CHEMISTRY

Date: 03-04-2019	Dept. No.	Max. : 100 Marks
Time: 09:00-12:00		

#### Part-A

#### Answer any FOUR questions.

 $(4 \times 10 = 40)$ 

- 1. a) Explain the synthesis of Schiff base macrocyclic metal complexes with an example.
  - b) Discuss whether the following complexes possess orbital contribution to the magnetic moment: d<sup>2</sup> (Oh) and d<sup>7</sup> (Td). (5+5)
- 2. Describe the possible electronic excitations of metals ions with  $d^2$  configuration in octahedral environment using Orgel diagram.
- 3. Explain the principle involved in the polarographic analysis of metal complexes.
- 4. Discuss any three evidences which support crystal field theory.
- 5. What are inert and labile complexes? Explain the effect of electronic configuration of d<sup>1</sup> to d<sup>9</sup>to predict whether the metal complexes are inert or labile.
- 6. Predict whether the following oxides are spinel or inverse spinel i) Mn<sub>3</sub>O<sub>4</sub> ii)ZnFe<sub>2</sub>O<sub>4</sub>.
- 7. Briefly explain the role of coordination compounds in *invivo* and *invitro* nitrogen fixation.
- 8. Explain the role of g- and A-tensors in characterizing the EPR spectrum of metal complexes.

### Part-B

## Answer any THREE questions.

 $(3 \times 20 = 60)$ 

- 9. a) Discuss the principle involved in explaining the ESR spectrum of metal complexes with  $[Cu(salen)_2]^{2+}$  as an example.
  - b) How does cyclic voltammentry help in analyzing the electrochemical reversibility of coordination compounds? (10+10)
- 10. Explain the principle involved in the analysis of metal complexes by the Mossbauer spectroscopy and explain how the metal complexes of Fe<sup>3+</sup> and Fe<sup>2+</sup> are differentiated.
- 11. Write a brief note on static and dynamic Jahn-Teller effect. Which d<sup>n</sup> configuration leads to weak and strong Jahn-Teller distortion in octahedral and tetrahedtal complexes?
- 12. a) Explain the mechanism and factors affecting the inner and outer sphere electron transfer reaction of metal complexes.
  - b) What is trans effect? Discuss its mechanism with suitable example. (10+10)
- 13. a) Discuss in detail the trans effect.
  - b) Discuss why is the trans effect of CO higher than that of pyridine. (10+10)
  - 14. Compare the specific role of the enzymes (i) carboxy peptidase and (ii) carbonic anhydrase. (10+10)

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