



LOYOLA COLLEGE (AUTONOMOUS), CHENNAI – 600 034

B.Sc. DEGREE EXAMINATION – CHEMISTRY

THIRD SEMESTER – APRIL 2019

CH 3506– ORGANIC FUNCTIONAL GROUPS - I

Date: 25-04-2019
Time: 01:00-04:00

Dept. No.

Max. : 100 Marks

PART- A

Answer ALL questions

10X2 = 20 marks

1. Give the IUPAC name of
(i). $\text{CH}_3\text{CH}_2\text{CHBrCH}_2\text{CHClCH}_3$ (ii). $\text{C}_6\text{H}_5\text{CH}_2\text{Cl}$
2. How does methane react with chlorine under sunlight?
3. What is epoxidation reaction? Give an example.
4. How will you convert acetone into propane.
5. Write the IUPAC name of (i). Ethyl methyl ether (ii), Methyl isopropyl ether
6. How will you prepare diethyl ether by Williamson's synthesis?
7. Differentiate ketone from aldehydes.
8. What is aldol condensation? Give an example.
9. State and explain esterification reaction with an example.
10. Formic acid is a stronger acid than acetic acid. Why?

PART- B

Answer any EIGHT questions

8x5 = 40 marks

11. Explain the mechanism of $\text{S}_{\text{N}}2$ reaction with an example.
12. Mention the applications of Hoffmann and Saytzeff rules.
13. Why is phenol acidic? Explain.
14. Convert the following :
 - (i). Acetaldehyde into Isopropylalcohol **3 marks**
 - (ii). 1- Butene into 2-Butanol **2 marks**
15. How does ethanol react with the following
 - (i). Sulphuric acid **3 marks**
 - (ii). Acetic acid **2 marks**
16. (i). Mention any three uses of diethyl ether **3 marks**
(ii). How does diethyl ether react with inner mineral acids. **2 marks**
17. Differentiate alcohols and ethers.
18. Give a brief account on Norrish type- I photo chemical reaction of carbonyl Compounds.

19. What is MPV reduction? Discuss its mechanism.
20. What is urotropine? How will you prepare it? Mention its use.
21. Complete the following
- (i). $\text{C}_2\text{H}_5\text{MgBr} \xrightarrow{\text{CO}_2, \text{H}^+} ?$
- (ii). $\text{C}_2\text{H}_5\text{COONa} + \text{NaOH} + \text{heat} \xrightarrow{\hspace{2cm}} ?$
- (iii). $\text{C}_2\text{H}_5\text{COOH} \xrightarrow{\text{LiAlH}_4} ?$
22. Explain alkaline hydrolysis of esters and trans esterification.

PART- C

Answer any FOUR questions

4x10 = 40 marks

23. (i). Explain the mechanism of E1 and E2 reactions with examples **6 marks**
 (ii). Discuss the role of nucleophile and solvent in nucleophilic substitution reaction.
4 marks
24. Explain the following reactions with mechanism **5 + 5 = 10 marks**
 (i). FriedelCraft's alkylation reaction (ii). Reimer – Tiemen reaction
25. How does diethyl ether react with the following : **5x2 = 10 marks**
 (i). PCl_5 (ii). CH_3COCl (iii). HI (iv). H_2SO_4 (v). Cl_2 / h
26. Write notes on the following reactions with mechanism
- (i). Perkin's reaction **3 marks**
 (ii). Benzoin condensation **3 marks**
 (iii). Cannizaro reaction **4 marks**
27. Starting from acetic acid how will you prepare **5x2 = 10 marks**
 (i). Trichloro acetic acid (ii). Acetyl chloride (iii). Acetamide
 (iv). Acetic anhydride (v). Methane
28. (i). What is Wittig reaction? Explain its mechanism. **5 marks**
 (ii). Discuss the action of heat on various hydroxy acids. **5 marks**
