ALTO ALTEONOMOUS), CERNNAI - 600 034

egree examplation - food chemistry & food frocessing

FIRST SEMESTER - NOVEMBER 2013

I'P 1806 - ORGANIC CHEMISTRY OF FOOD - I

Date: 05/11/2013 Dept. No. Max.: 100 Marks
Time: 1:00 - 4:00

Part-A

Answer all the questions. Each question carries two marks.

- Distinguish between asymmetric and dissymmetric molecules with examples.
- Explain the structural features of glucose which make it soluble in water.
- What are the structural features of carrageenan?
- 4. Why are dietary fibers necessary in food? Mention any two natural sources rich in dietary fibers.
- 5. Explain the chemical reaction involved in the hydrogenation of fat with an example.
- 6. Explain why unsaturated fats are more favourable for consumption than saturated fats.
- 7. Define isoelectric point.
- 8. What is the biological function of enzymes?
- 9. Explain the chemical reaction responsible for the swelling of starch.
- 10. Mention any two naturally occurring sources of ascorbic acid.

Part-B

Answer eight questions. Each question carries five marks.

- 11. Explain the chemical reaction involved in the hydrolysis of a polysaccharide with a suitable example.
- 12. What are the products resulting from the chemical oxidation and chemical reduction of a monosaccharide? Give suitable examples.
- 13. Explain the Somoyogi-Nelson method for estimation of total reducing sugars.
- 14. How can the total starch content in pectins be estimated?
- 15. Discuss any one chemical method for the evaluation of protein content in food.
- 16. Explain the chemical reaction of amino acids with ninhydrin.
- 17. Write the structural formulae of glycylalanine and alanylglycine.
- 18. Discuss the determination of vitamin A using the HPLC method.
- 19. What are the structural features that cause hydrophobic interactions in proteins?
- 20. Explain the HPLC method of determining vitamin E.
- 21. Describe how disulphide bonds are formed in amino acidswith an example.
- 22. Discuss the oxidation reactions of L-ascorbic acid with reference to the effect of pH on the same.

Part-C

Answer four questions. Each question carries ten marks.

- 23. Discuss the factors affecting the reactions of enzymes.
- 24a. Write the structure of pantothenic acid, mention its sources and explain its biological role.
 - b. What are emulsions? Describe a method of emulsification.

(5+5)

- 25. Discuss the forces involved in the stability of the structure of a protein.
- 26. What is the mechanism of catalysis by enzymes and derive the equation for the same?
- 27. Explain the significance of the chemical reactions in proteins with respect to food processing. What are the changes in the physical properties as a consequence of such reactions?
- 28a. Explain the factors influencing the consistency of commercial fats.
 - b. Explain the use of enzymes in improving nutritional quality in food with suitable examples. (5+5)
