

**LOYOLA COLLEGE (AUTONOMOUS) CHENNAI – 600 034****M.Sc. DEGREE EXAMINATION – CHEMISTRY****SECOND SEMESTER – NOVEMBER 2024****PCH2MC01 – ORGANIC REACTION MECH. AND HETEROCYCLIC COMPOUNDS**

Date: 11-11-2024

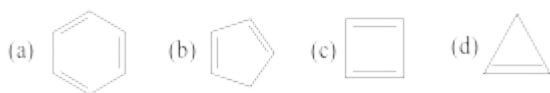
Dept. No. 

Max. : 100 Marks

Time: 01:00 pm-04:00 pm

**SECTION A – K1 (CO1)****Answer ALL the questions****(5 x 1 = 5)****1. Multiple choice questions**

a) Which of the following compounds is anti-aromatic?



b) The number of ESR lines given by a methyl radical are

(a) 4 (b) 5 (c) 6 (d) 7

c) Which of the following is not involved in Rosenmund reduction?

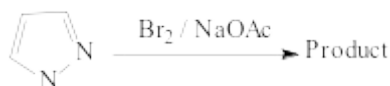
(a) Acid chloride (b)  $H_2$  (c)  $Pd-BaSO_4$  (d) Aluminium isopropoxide

d) The compound/s which does not undergo Markovnikov's addition is/are \_\_\_\_\_.

(i)  $CH_2=C=CH_2$ , (ii)  $CH_2=C=O$ , (iii)  $CH_2=CH-CF_3$ 

(a) (iii) only (b) (ii) &amp; (iii) (c) (ii) only (d) all of these

e) Predict the substitution position for bromination in the following reaction.

(a) 3<sup>rd</sup> position (b) 4<sup>th</sup> position (c) 5<sup>th</sup> position (d) 2<sup>nd</sup> position**SECTION A – K2 (CO1)****Answer ALL the questions****(5 x 1 = 5)****2. Multiple choice questions**

a) Which of the following statement is incorrect?

- (a) Anti-aromatic compounds are more stable than non-aromatic compounds.  
(b) Cyclooctatetraene is a tub-shaped molecule.  
(c) Homoaromatic compounds contain a  $sp^3$  hybridized carbon.  
(d) Aromatic compounds obey Huckel's  $4n+2$  rule.

b) The intermediate in  $S_N1$  reaction is a

(a) radical (b) carbanion (c) carbocation (d) carbene

c) Which of the following statement/s is/are correct?

- (a)  $E1$  reaction follows Saytzeff rule.  
(b)  $E2$  reaction never gives Hofmann product.  
(c)  $E1cb$  reaction occur under acidic conditions.  
(d) All of these

d) Benzyne + anthracene forms

(a) triptycene (b) naphthalene (c) phenanthrene (d) all of these

e) Imidazole on reaction with nitrating mixture forms

	(a) 4-nitroimidazole      (b) 2-nitroimidazole      (c) 5-nitroimidazole      (d) none
<b>SECTION B – K3 (CO2)</b>	
	<b>Answer any THREE of the following</b> <span style="float: right;"><b>(3 x 10 = 30)</b></span>
3.	(a) Compare the conditions for aromatic, non-aromatic and anti-aromatic nature of compounds. (5) (b) Explain the mechanism and limitations of Friedel Crafts acylation. (5)
4.	(a) Which of the following will undergo solvolysis more readily and why? (5) $(\text{C}_6\text{H}_5)_2\text{CHBr}$ or $(\text{CH}_3)_3\text{CBr}$ (b) Explain $\text{S}_{\text{N}}\text{Ar}$ reaction mechanism with suitable evidences. (5)
5.	Discuss the mechanism and evidences of $\text{E1}$ reaction.
6.	(a) Predict the product for the reaction of cyclohexenone with Gilman reagent. (4) (b) Why are alkynes less reactive towards addition reaction than alkenes? (3) (c) "Wittig reagents are carbon nucleophiles." Justify this statement. (3)
7.	How are 1,3-oxazole and thiazoles synthesised by Hantzsch method? Discuss the substitution position with an electrophile? (10)
<b>SECTION C – K4 (CO3)</b>	
	<b>Answer any TWO of the following</b> <span style="float: right;"><b>(2 x 12.5 = 25)</b></span>
8.	(a) Discuss the orientation and reactivity of nitrobenzene. (6) (b) Write the mechanism and uses of Stark-enamine reaction. (6.5)
9.	(a) Explain the Gomberg method of the formation of triphenyl methyl radical. (6.5) (b) How would you detect a long-lived and a short-lived free-radicals? Explain. (6)
10.	(a) Account for the following statements: (i) Stereospecificity of bromination reaction of $\text{PhCH}=\text{CHCH}_3$ decreases in solvents of high dielectric constant. (4) (ii) Addition of $\text{Br}_2$ in ethylene in methanol solvent is retarded by the addition of $\text{KBr}$ and results in the formation of a mixture of $\text{Br}-\text{CH}_2-\text{CH}_2-\text{Br}$ and $\text{Br}-\text{CH}_2-\text{CH}_2-\text{OCH}_3$ . (4) (b) How is 1,3-oxazole synthesised by Robinson-Gabriel synthesis? (4.5)
11.	(a) Identify A & B and justify your answer with suitable mechanism. (4+4) <div style="text-align: center;"> <p>The reaction scheme shows a cyclohexenone ring with a vinyl group at the 2-position. The vinyl group is attached to a <math>\text{C}_6\text{H}_5</math> group. The reaction with <math>\text{H}_2\text{O}_2/\text{AcOH}</math> yields product A, and the reaction with <math>\text{H}_2\text{O}_2/\text{NaOH}</math> yields product B.</p> </div>
	(b) What is Grunwald-Winstein relationship? (4.5)
<b>SECTION D – K5 (CO4)</b>	
	<b>Answer any ONE of the following</b> <span style="float: right;"><b>(1 x 15 = 15)</b></span>
12.	(a) Explain the aromaticity of [10]-, [14]-, and [18]-annulenes. (9) (b) Explain single electron transfer mechanism (SET) with evidences. (6)
13.	(a) 'The effect of attacking nucleophile in $\text{S}_{\text{N}}1$ reaction kinetics is negligible.' Why? (5) (b) Explain Sommelet-Hauser rearrangement with mechanism. (5) (c) Draw the resonance structures of 1,2,3-triazole and tetrazole and explain their reactivity towards electrophilic substitution reactions. (5)
<b>SECTION E – K6 (CO5)</b>	
	<b>Answer any ONE of the following</b> <span style="float: right;"><b>(1 x 20 = 20)</b></span>
14.	(a) Illustrate the mechanism, evidences, orientation and reactivity of $\text{E2}$ reaction. (12) (b) Give the evidences for the syn, anti, and non-stereoselective electrophilic addition of olefins. (8)
15.	(a) Predict the product for the addition of carbenes to <i>trans</i> -2-butene in gaseous and liquid phases. (5)

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|  | (b) Identify the position of an electrophilic substitution in the 1,2-azoles and write the nitration, halogenation reactions on pyrazoles and isothiazoles. (10) |
|  | (c) Write any one method of synthesis for pyridazine. (5)  |

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