## LOYOLA COLLEGE (AUTONOMOUS), CHENNAI – 600 034



## **B.Sc.** DEGREE EXAMINATION – **PHYSICS**

## SIXTH SEMESTER - NOVEMBER 2016

PH 6611 - ATOMICS AND NUCLEAR PHYSICS	
Date: 14-11-2016 Dept. No. Time: 09:00-12:00	Max.: 100 Marks
PART-A Answer ALL questions	(10 x 2 =20 marks)
<ol> <li>Write the properties of positive rays.</li> <li>Define Pauli's exclusion principle.</li> <li>Explain fine structure of the sodium D-line.</li> <li>Calculate the wavelength separation between the unmodified line of wavelength 6000 Å and the modified lines when a magnetic induction of 1 Wbm<sup>-2</sup> is applied, in normal Zeeman effect.</li> <li>Classify isobar and isotope with example.</li> <li>Define range of an α particle.</li> <li>Define chain reaction.</li> <li>Explain nuclear fission. Give example.</li> <li>State Hubble's law.</li> <li>What are cosmic ray showers?</li> </ol>	
PART –B Answer any FOUR questions	(4 x 7.5 = 30 marks)
<ol> <li>What are positive rays? Describe Thomson's method for positive ray analysis.</li> <li>Explain about (i) L-S coupling and (ii) j-j coupling.</li> <li>Explain (i) mass defect (ii) binding energy (iii) packing fraction.</li> <li>Discuss the liquid drop model of a nucleus.</li> <li>Give an account of anti-particles with suitable examples.</li> <li>Explain the four fundamental interactions in elementary particles.</li> </ol>	(7.5) (4+3.5) (3x2.5) (7.5) (7.5)
PART – C	(1.5)
Answer any FOUR questions (4)	x12.5=50 marks)
7. Give an account of Bohr – Sommerfield model of an elliptical electron orbits of l How does it account for the fine structure of hydrogen atom?	(12.5)
8. Describe the experimental arrangement of Zeeman Effect and Anomalous Zeema	an effect. (12.5)
<ul> <li>19. Describe the method of measuring the range of α particle using Geiger and Nutta</li> <li>20. Discuss in detail the Gamow's theory of α-decay.</li> <li>21. (i) Distinguish between primary and secondary cosmic rays. (ii) Write a note on</li> </ul>	(12.5) (12.5)
variation with latitude, altitude, longitude, and east-west direction. 22. Explain in detail about the quantum numbers in elementary particle.	(2.5+10) (12.5)